NYC HIGH SCHOOL SCIENCE REGENTS SCOPE & SEQUENCE



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CHEMISTRY

For the purposes of this document, a period is defined as a 45-minute block of time and is meant to be for planning only.

| first term: september – january |

UNIT 1	UNIT 2	UNIT 3	UNIT 4	UNIT 5
The Physical Nature	Atomic Concepts	Nuclear Chemistry	Chemical Bonding	Periodicity
of Matter (30 days)	(20 days)	(5 days)	(20 days)	(5 days)
 Definition of Chemistry Matter 3.1dd, Particulate 3.1nn,	• Atoms	 Radioactivity 3.1m, 3.1n, 3.1o, 3.1p, 4.4d, 4.4e, 4.4f Nuclear Energy 4.1a, 4.4a, 4.4b, 4.4c Alternative 4.4d, Energy Sources 4.4e, 4.4f, 5.3a, 5.3b, 5.3c Medical Applications 4.4d 	 Why Atoms Bond 5.2b, 5.2c, 5.2i Bonds 5.2a, 5.2b, Between 5.2c, 5.2d, Atoms 5.2e, 5.2g, 5.2h, 5.2i, 5.2k Molecular Attraction 5.2j, 5.2l, 5.2k Solubility 3.1oo, 3.1pp, 3.1qq Chemical Formula 3.1cc, 5.2g, 3.3e Chemical Equation 3.2b, 3.3a, 3.3c 	 Development of the Periodic Table Properties and Trends of the Elements 3.1bb. Elements 3.1y, 3.1l, 3.1v, 3.1x, 3.1y, 3.1z, 5.2b, 5.2f, 5.2f, 5.2h

The Reference Tables for the Chemistry document is an integral component of the Regents Chemistry course and can be found at: www.emsc.nysed.gov/osa/scire/reftable.html Scientific Inquiry (i.e., asking questions, making discoveries, gathering data, analyzing explanations and communication) is an integral component of this course.

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CHEMISTRY

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second term: february - june |

UNIT Moles/Stoich (12 day	niometry	UNIT Kinetics and Ed	quilibrium	UNIT 8 Acids and Bases (15 days)		UNIT Carbon and Chemistry	d Organic	UNIT 10 Review (10 days)
 Moles/ Stoichiometry Mole Interpretation	3.1cc, 3.1ee, 3.1n, 3.3e, 3.4e	 Kinetics Equilibrium 	3.4d, 3.4f, 3.4a, 3.4b, 3.4c, 3.4g	 Electrolytes Acids and Bases Acid-Base Reactions 	3.1rr, 5.2n 3.1uu, 3.1vv, 3.1ww, 3.1yy, 3.1ss, 3.1tt 3.1xx, 3.1zz, 3.2b	 Carbon Chemistry Classify, Name and Identify Organic 	3.1ff, 3.1gg, 3.1hh, 3.1ii 3.1ff, 3.1gg, 3.1hh, 5.2e	First-Term TopicsRegents Exam Prep
Use of the Mole Concept	3.3a, 3.1oo, 3.1pp, 3.1qq	3.4n, 3.4i, 3.4j, 4.1c, 4.1d,		-Reduction days)	Compounds Based on Structure, Bond Type			
Stoichiometry	3.3f, 3.2b		4.2b, 4.2c, 3.1ll, 3.1mm	• Redox	3.1b, 3.2d, 3.2e, 3.2f, 3.2g, 3.2h, 3.3b, 3.2i	• Properties of Organic Compounds	3.1hh, 3.1ii, 5.2e	
Solutions	3.1oo, 3.1pp, 3.1qq			Electro- chemistry	3.3a, 3.1i, 3.2j, 3.2k, 3.2l	Reactions	3.2c	

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EARTH SCIENCE

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first term: september – january

UNIT 1 Maps and Measurements (17 periods)	and Measurements Dynamic Earth		UNIT 4 Landscapes (30 periods)	UNIT 5 Earth History (12 periods)	
 Short Introduction of Origin of Earth 1.2a, and Our Place in the Universe (Big Bang, 1.2b Solar System) Measurements, Reference Tables, Graphing, Nature of Earth Science Locating Points on the Earth, Latitude, 1.1d, 	 Structure of Earth and Properties Convection Cycles and Density Evidence of Movement Plate Tectonics 2.1a, 2.1j 2.1a, 2.1b 2.1j, 2.1k 2.1l, 2.1m 	 Minerals 3.1a, 3.1b, 3.1c Igneous Rocks 3.1c Metamorphic Rocks 3.1c Sedimentary 1.2.f, 	 Water cycle 1.2g, 2.1b, 2.1u Hydrology 1.1i, 1.2f, (Stream 1.2g, 2.1r, Mechanics, 2.1u, 2.1v, Ground Water) 2.1w Weathering 1.2g, 2.1s, Agents 2.1t, 2.1u, 3.1c Erosion and Deposition 2.1r, 2.1s, 2.1s, 2.1s, 2.1s, 2.1s 	 Fossils 1.2d, 1.2f, 1.2h, 1.2i, 1.2j Geologic Time 1.2h, 1.2i, 1.2i, 1.2 Stratigraphy 1.2j, 2.1o 	
Longitude, Maps 1.1e, 1.1g, 2.1q • Isomaps (Topographic Maps) • GPS/GIS	2.11, 2.1m, 2.1n, 2.1o • Earthquakes and Volcanoes 2.1j, 2.1k, 2.11, 2.1m, 2.1n 2.1n	Rocks (Intro— 2.1v, May Be Taught 2.1w, with Weathering) 3.1c • Mining and 3.1a, Natural Resources 3.1c	2.1t, 2.1u, 2.1w, 2.1v • Sedimentary Rocks If Not Covered Previously • Soils (Porosity, Permeability) 2.1u • Real-World 2.1p, 2.1t, Applications: 2.1u, 3.1c Agriculture, Mud Slides	• Radioactive 1.2j Dating	

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EARTH SCIENCE

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second term: february - june |

UNIT 6	UNIT 7	UNIT 8	UNIT 9	UNIT 10
Insolation	Meteorology	Climate	Astronomy	Review
(13 periods)	(17 periods)	(10 periods)	(17 periods)	(10 days)
 Arc of Sun's Travel 1.1h, 2.2a Seasons 1.1a, 1.1f, 1.1h Energy 1.2e, Exchanges in the Atmosphere 2.1i, 2.2a, 2.2b 	 Systems	• Factors that Affect Climate (Altitude, 2.1i, Latitude) 2.1o, 2.1r, 2.1s, 2.2b, 2.2c • Water Budget (Concept—Not Actual Budget) 1.2g, 2.2c	 Phases of the Moon 1.1f, 1.1i Solar System— Eccentricity 1.1a Tides 1.1i Celestial 1.1a, Observations, 1.1f, HR Diagram 1.1g, 1.2a, 1.2b 	 First-Term Topics Regents Exam Prep

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LIVING ENVIRONMENT

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| first term: september – january |

UNIT 1	UNIT 2	UNIT 3	UNIT 4 Organization and Patterns in Life (20 periods)	UNIT 5
Scientific Inquiry	Origin of Life	Ecology		Homeostasis and
(10 periods)	(3 periods)	(22 periods)		Immunity (25 periods)
 The Role of Scientific 1.1a, 1.1b, 1.1c, 3.1, 3.2, 3.3 Biology The Methods of Science Standard 1– 1.2a, 1.2b, 1.3a, 1.3b, 2.1, 2.2, 2.3a, 2.3b, 2.3c, 2.4, 3.4a, 3.4b, 3.4c, 3.5a, 3.5b 	Formation of First Cells from Molecules The Nature of Prokaryotes 1.3a, 3.1j	• Relationships 1.1c, 1.1d, 6.1g, 6.2a, 6.2b, 6.3a • Interactions 1.1a, 1.1b, 1.1d, 1.1e, 1.1f, 6.1a, 6.1b, 6.1c, 6.1d, 6.1e, 6.1f, 6.3b, 6.3c	 Cell Structure 1.2a, 1.2e, 1.2g, 1.2i, Cell Physiology 1.2c, 1.2f, 1.2g, 1.2i, 1.3a Cell Chemistry 1.2h, 1.2j, 5.1c, 5.1f, 5.1g Photosynthesis 5.1a, 5.1b Respiration 5.1d, 5.1e, 5.1f Diffusion and Osmosis 4.1a, 4.1b 	 Body System Overview

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LIVING ENVIRONMENT

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second term: february - june

UNIT 6 Reproduction and Development (30 periods)	UNIT 7 Genetics and Biotechnology (25 periods)	UNIT 8 Evolution (15 periods)	UNIT 9 Human Influences on the Environment (15 periods)	UNIT 10 Review (10 periods)
 Meiosis 4.1c Reproductive Systems 4.1a, 4.1e, 4.1f, 4.1g Fertilization 2.1e, 4.1c Development 4.1d, 4.1e, 4.1h Stem Cells 	 Mendel Overview Core Curriculum LE 2.2a, 2.2b, 2.2c DNA/RNA 2.1a, 2.1b, 2.1c, 2.1f Protein 2.1g, 2.1i, 2.1j, 2.1k Diseases 2.2e, 5.2h Mutations 2.1h, 2.2d, 2.2e, 5.2i Bio-engineering 2.2a, 2.2b, 2.2c Bioethics 1.2c 	• Natural 3.1a, Selection 3.1b, 3.1c, 3.1d, 3.1e, 3.1f, 3.1g, 3.1h, 3.1i, 3.1j, 3.1k, 6.2a • Evidence 3.11	 Positive T.1a, Influences 7.1b Negative T.2a, 7.2b, 7.2c Decision T.3a, Making T.3b (Risk/Benefit) 	First-Term Topics Regents Exam Prep

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PHYSICS

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first term: september – january |

UNIT 1 Measurement and Mathematics (15 days)	UNIT 2 Mechanics (40 days)		UNIT 3 Energy (15 days)		UNIT 4 Projects and Problem Based Learning Activities (10 days)	
 Units Standard 1-M1.1 SI Prefixes Tools in Measurement Scientific Standard 6-3.2 Significant Figures PS 5.2c Evaluating Experimental Results Graphing Data M1.1, 2.1 Scalar and Vector Quantities Solving M1.1 	 Kinematics Free Fall: Gravity Distance/Displacement Speed/Velocity Acceleration Statics Forces Dynamics Newton's Three Laws of Motion Two-Dimensional Motion and Trajectories: Fired Horizontally and at an Angle Uniform Circular Motion Newton's Universal Law of Gravitation Gravitational Field Strength Weight Friction Incline Plane Momentum	5.1d 5.1b, 5.1c, 5.1 j 5.1e, 5.1i, 5.1k, 5.1q Standard 6– 4.2 5.1b, 5.1c, 5.1f, 5.1g, 5.1h 5.1n 5.11, 5.1n, 5.1s, 5.1t, 5.1u 4.1h, 5.1o	 Work and Energy Power Forms of Energy Potential Energy Elastic Potential Energy Hooke's Law Kinetic Energy Work-Energy Relationship Conservation 	4.1g, 4.1h, 4.1j 4.1a, 4.1i, 5.3f, 5.3j 4.1c 4.1c, 5.1m 4.1d 4.1a, 4.1b, 4.1e, 4.1f,	 Egg Drop Trebuchet Bridge Building 	Standard 1– S2.1, 2.2, 2.3, 2.4 Standard 2– 1.1-1.5
Equations Using Algebra	Law of Conservation The Simple Pendulum	5.1p, 5.1r 4.1a, 4.1c, 4.1d, 4.1e		4.1j		

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PHYSICS

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second term: february - june

UNIT 5 Electricity and Magnetism (25 days)		UNIT 6 Waves (25 days)		UNIT 7 Modern Physics (20 days)		UNIT 8 Review (10 days)
 Electrostatics Structure of Atoms Charged Objects Transfer of Charge Law of Conservation Coulomb's Law Electric Fields Field Strengths Potential Difference Electric Currents Ohm's Law Resistivity Electric Circuits Series and Parallel Electric Power 	5.1t, 5.3b, 5.3f 5.1s, 5.1u 4.11, 4.1n 4.1a, 4.1b, 4.1j, 4.11, 4.1m, 4.1n, 4.1o, 4.1p	 Introduction to Waves Types of Waves Wave Behavior Wave Characteristics Pulses and Periodic Wave Periodic Wave Phenomena Doppler Effect Interference Standing Waves Resonance Diffraction Sound Light Speed of light Reflection Refraction 	4.3a, 4.3b 4.3c, 4.3h 4.3d, 4.3e, 4.3f, 4.3m, 4.3n, 4.3h, 4.3i, 4.3j 4.3l, 4.3i, 4.3j, 4.3h, 4.3k	Wave-Particle Duality of Energy and Matter Quantum Theory Energy vs. Frequency Photon-Particle Collisions (DeBroglie & Compton) Early Models of the atom Thomson's Model Rutherford's Model Bohr's Model Atomic Spectra Cloud Model The nucleus Nuclear Force Universal Mass Unit Mass-Energy	5.3e 5.3a, 5.3c, 5.3d	First-Term Topics Regents Exam Prep
 Electric Energy Magnetism Fields, Flux Lines, and Strength Electromagnetic Induction Electromagnetic Radiation 	5.1t	 Absolute Index of Refraction Snell's Law The Electromagnetic Spectrum Frequencies and Wavelengths 	4.1b, 4.1k, 4.3g, 4.3k	Relationship • The Standard Model of Particle Physics – Fundamental Forces – Classification of Subatomic Particles – Quarks	4.1b, 5.3b, 5.3f, 5.3g, 5.3i, 5.3j	

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acknowledgements

NYC Science Scope and Sequence High School was produced under the auspices of Joel I. Klein, Chancellor and Dr. Marcia V. Lyles, Deputy Chancellor for Teaching and Learning. Grateful acknowledgement is extended to the following people for their valuable contribution to the development of this document:

Department of Mathematics and Science: Linda Curtis-Bey, Director of Mathematics and Science; Denise McNamara, Science Instructional Specialist.

Judith Abel, RIS; Orlando Avila, High School for Medical Sciences; Nadya Awadallah, RIS; Robert Kane, RIS; Mindy Molina, Jane Addams High School; Marc Siciliano, Stevenson Campus; Marianita Damari, RIS; Susan Brustein, Townsend Harris High School; Najla Hallak, Martin Van Buren High School; Ellen Mandel, MAST Magnet; Michael Renna, Hillcrest High School; Robert Soel, John Bowne High School; Michael Cambria, RIS; John Pritchard, Grover Cleveland High School; Vladimir Hurych, Long Island City High School; Diane Pillersdorf, RIS; Gregory Arnold, Beach Channel High School; Norm Cohn, Franklin K. Lane High School; Patrice English, John Adams High School; Mary Lanagan, Transit Tech High School; Christine Mineo, RIS; Douglas Graham, Erasmus-Humanities High School; George P. Matthew, George Wingate High School; Harold Meiselman, Clara Barton High School; Michael McDonnell, Midwood High School; Henrietta Staub, L.M. Goldstein High School; Joseph Scarmato, RIS; Alan Asher, Port Richmond High School; Gina Battista, Tottenville High School; Traci Frey, Edward R. Murrow High School; Douglas Friend, Edward R. Murrow High School; Sylvia Goldberg, Abraham Lincoln High School; Elizabeth Wozniak Towlen, Edward R. Murrow High School; Derresa Davis, RIS; Eric Megli, Boys and Girls High School; Karen Hutchinson, Boys and Girls High School; Raju Philip, George Westinghouse High School; Tarik Zarrouk, ACORN HSSJ; Megan Roberts, RIS; Theresa Budney, Marte Valle, Ingrid Buntschuh, Seward Park High School; John Gollisz, High School for Environmental Studies; Elizabeth Fong, Stuyvesant High School; James Johnson, High School for Art & Design; Olga Livanis, Stuyvesant High School; Jose Merced, John Dewey High School; Judy Silverman, High School of Graphic Design; Tammy Vu, East Side Community High School; Sheldon Young, RIS; William Dugan, High School for Mathematics, Science and Engineering @ CUNY; Barbara Poseluzny, A. Philip Randolph High School; Julie Torres, J. Levine High School for Media and Communications; Craig Weiss, La Guardia High School for Performing Arts.

Special Acknowledgement goes to Dr. Julia Rankin and Marion Zachowski for their invaluable efforts in helping to create this document.